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## FORMATION OF OXOBORANE AND THIOXOBORANE FROM A DITHIASTANNABORETANE DERIVATIVE

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Reaction with an oxygen-donating reagent such as DMSO and thermolysis of a 1,3,2,4-dithiastannaboretane derivative bearing 2,4,6-tris[bis(trimethylsilyl)methyl]phenyl (Tbt) group led to the formation of novel boron-group 16 element double bond compounds, oxoborane (Tbt-B=O) and thioxoborane (Tbt-B=S). The oxoborane and thioxoborane underwent cycloaddition reactions to give the corresponding adducts in good yields.

<u>Keywords</u>: 1,3,2,4-dithiastannaboretane; thermolysis; boroncontaining doubly-bonded compounds; oxoborane; thioxoborane; cycloaddition

#### INTRODUCTION

In contrast to the extensive studies on the stable methyleneboranes (RB=CR<sub>2</sub>)<sup>[1]</sup> and iminoboranes (RB=NR),<sup>[2]</sup> very little is known for the chemistry of boron-group 16 element double-bond compounds.<sup>[3]</sup> We previously reported the synthesis of the first stable dimercaptoborane 1 bearing a 2,4,6-tris[bis(trimethylsilyl)methyl]-phenyl (Tbt) group and its facile transformation to the novel four-

membered boracycles 2–5.<sup>[4, 5]</sup> Here, we present some reactions of the tin-containing four-membered boracycle 4 leading to the formation of a kinetically stabilized boron-oxygen double-bond compound (oxoborane) and its sulfur analogue (thioxoborane).

$$\begin{array}{c|c} Me_3Si & H & H & SiMe_3 \\ Me_3Si & SiMe_3 & \equiv Tbt \\ & & SiMe_3 \\ & & SiMe_3 \end{array}$$

#### RESULTS AND DISCUSSION

#### Synthesis and Structure of 1,3,2,4-Dithiametallaboretanes

Dilithiation of the dimercaptoborane, TbtB(SH)<sub>2</sub> (1), which was synthesized by the sulfurization of the corresponding overcrowded lithium aryltrihydroborate bearing 2,4,6-tris[bis(trimethylsilyl)methyl]-phenyl (Tbt) group, followed by treatment with electrophiles such as Cp<sub>2</sub>TiCl<sub>2</sub>, Mes<sub>2</sub>GeBr<sub>2</sub>, Ph<sub>2</sub>SnCl<sub>2</sub>, and TbtSbBr<sub>2</sub> resulted in the isolation of novel four-membered boracycles, 1,3,2,4-dithiametallaboretanes 2-5 as stable crystals, respectively.

TbtB(SH)<sub>2</sub> 
$$\frac{1. n\text{-BuLi}}{2. \text{ ECl}_2}$$
 Tbt-B S  $\frac{2}{3}$ ; E = GeMes<sub>2</sub> 4; E = SnPh<sub>2</sub> 5; E = SbTbt

ORTEP drawing of 4

#### Formation and Reactions of Oxoborane (Tbt-B=O)[6]

The dithiastannaboretane 4 was found to be a good precursor for the oxoborane 6. The reaction of 4 with dimethyl sulfoxide followed by the reaction with methanol and Mes\*CNO (Mes\* = 2,4,6-tri-t-butylphenyl) resulted in the formation of the expected methanol adduct 7 and [2+3]cycloadduct 8 of the oxoborane 6. Exposure of 6 to moisture gave a hydrolysis product 9.

#### Formation and Reactions of Thioxoborane (Tbt-B=S)[7]

The 1,3,2,4-dithiastannaboretane derivative 4 was found to undergo thermal retrocyclization to give a novel boron-sulfur double-bond compound (thioxoborane) 10, the formation of which was confirmed by trapping reactions with 1,3-dienes giving 11. Thioxoborane 10 was also generated by the thermal retro[4+2]cycloaddition reactions of the diene adduct 11.

4 
$$\xrightarrow{\Delta}$$
 Tbt-B=S  $\xrightarrow{R^1 R^2}$  Tbt-B  $\xrightarrow{R^1}$  R<sup>1</sup>

$$\xrightarrow{10}$$
  $\xrightarrow{R^1 R^2}$   $\xrightarrow{R^2}$   $\xrightarrow{\Delta/\text{retro } [4+2] \text{ cycloaddition}}$ 

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